

Absorbancia blanco= 0.2230

Absorbancia desconocido= 0.078

$$\text{mg\%CO} = \frac{0.2230 - 0.078}{0.2230} \times 27.75 = 18.04 \text{ mg\%CO}$$

$$18.04 \text{ mg\%CO} \times 0.8 = 14.4 \text{ ml\%CO}$$

$$\begin{array}{l} 1\text{g Hb} \quad \text{_____} \quad 1.34 \text{ ml CO} \\ 14\text{g Hb} \quad \text{_____} \quad x = 18.76 \text{ ml CO} \end{array}$$

18.76 ml CO _____ 100% Sat Hb (%COHb)

14.4 ml CO _____ X = 76.8 % Sat = 77% Sat Hb/ 77% COHb presente en la muestra